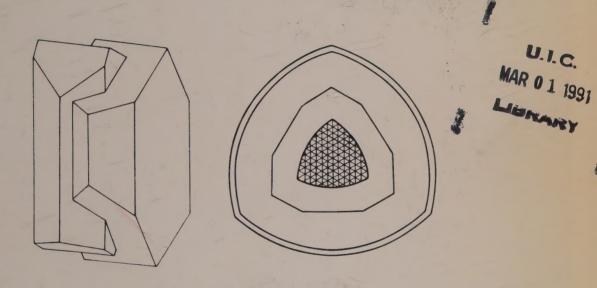
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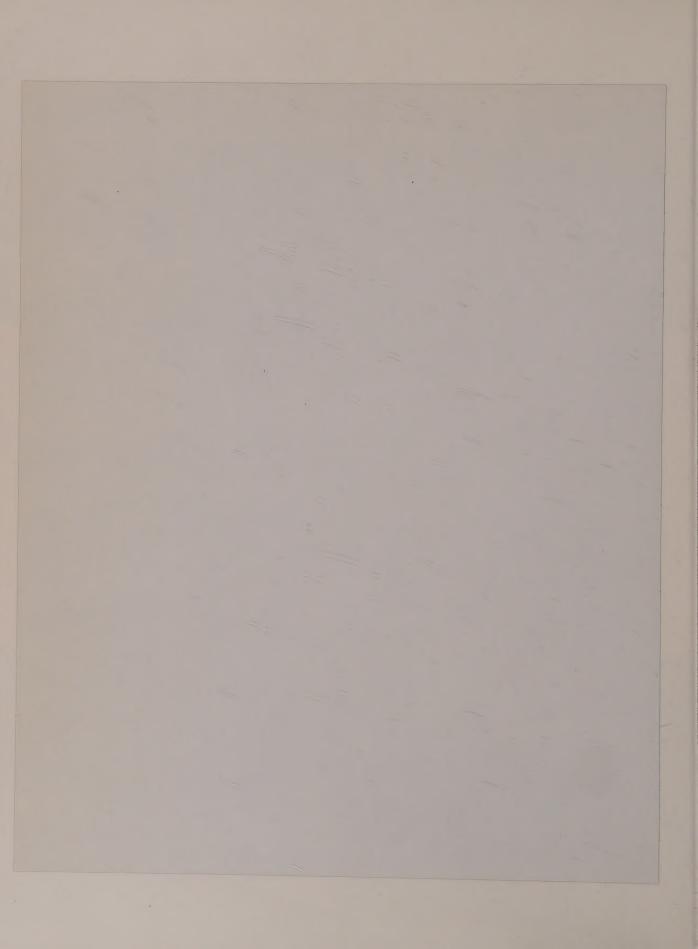
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DR. B. LINDQVIST, Naturhistoriska Riksmuseet, 104 05 Stockholm 50. Sweden: PD. DR. W. B. STERN, Mineralog.-Petrograph. Institut der Universität, Basel. Switzerland:

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ERRATA

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	for Na20 read Na2O	87M/3131	for Betakhtinde read Betekhtinite
	for St Helens read St Helena	87M/3134	for vysotaskite read vysotskite
87M/1734	for N. Mumayun read M. Humayun		for Machós read Machów
87M/3126	for $\dot{a} = a \equiv 2 \text{ read } \dot{a} = a \times 2$		for curces read curves
87M/3127	for meteoritic read meteoric	87M/3167	for γ 1.60, $2V_{\alpha}$ read γ 1.650, $2V_{\gamma}$
87M/3130	for chromite-rich read chrome-rich		y = 1 = ==, = + a + = a = 1 = = = 1

ORGANIZATION OF ABSTRACTS

Arising from a decision taken at the meeting of the INTERNATIONAL MINERALOGICAL ASSOCIATION in Copenhagen in 1961 the Mineralogical Societies of America and Great Britain agreed to issue a joint statement to National Societies adhering to the Association inviting each Society to organize contributions of abstracts of papers published in the journals of its country on subjects relevant to *Mineralogical Abstracts*. This invitation was issued and has brought a gratifying response. Members of Societies which have agreed to co-operate in this way are entitled to receive *Mineralogical Abstracts* for their personal use at a reduced rate of subscription on application, which must be made through their National Society. The countries now co-operating include: AUSTRALIA, AUSTRIA, BELGIUM, BULGARIA, CANADA, CZECHOSLOVAKIA, DENMARK, FINLAND, FRANCE, GERMANY, INDIA, ISRAEL, ITALY, JAPAN, NETHERLANDS, NEW ZEALAND, NORWAY, PAKISTAN, PORTUGAL, SOUTH AFRICA, SPAIN, SWEDEN, SWITZERLAND.

ABSTRACTORS

Contributors to this volume of *Mineralogical Abstracts* are:

Agrell, J. (J.A.), Gt. Britain; Aires Barros, L. (L.A.B.), Portugal; Akizuki, M. (M.Ak.), Japan; Alabaster, C. J. (C.J.A.), Gt. Britain; Arnaudova, R. (R.A.), Bulgaria; Aslanyan, S. (S.A.), Bulgaria; Atanasov, V. (V.A.), Bulgaria; Ball, D. F. (D.F.B.), Gt. Britain; Bass, M.A. (M.A.B.), Gt. Britain; Bayliss, P. (P.B.), Canada; Bennell-Baker, M. J. (M.J.B.-B.), Gt. Britain; Briggs, R. M. (R.M.B.), New Zealand; Browne, P. (P.Br.), Gt. Britain; Browne, P. R. L. (P.R.L.B.), New Zealand; Butt. K. A. (K.A.B.), Pakistan; Chisholm, J. E. (J.E.C.), Gt. Britain; Clayton, K. M. (K.M.C.), Gt. Britain; Clark, A. M. (A.M.C.), Gt. Britain; Coleman, L. C. (L.C.C.), Canada; Collier, H.A. (H.A.C.), U.S.A.; Colvine, R. J. L. (R.J.L.C.), Gt. Britain; Cooke, P. (P.C.), Gt. Britain; Cooper, J. W. (J.W.C.), Gt. Britain; Corsini, F. (F.C.), Italy; Crawford, M. L. (M.L.C.), U.S.A.; Cruikshank, A. R. I. (A.R.I.C.), Gt. Britain; Davis, D. J. (D.J.D.), Gt. Britain; Dietrich, R. V. (R.V.D.), U.S.A.; Dimmock, G. M. (G.M.D.), Australia; Donaldson, C. H. (C.H.D.), Gt. Britain;

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ABBREVIATIONS AND SYMBOLS

used in the text of abstracts

M.M Mineralogical Magazine	: M.A Mi	neralogical Abstracts : A.M Americ	can Mineralogist
CHEMICAL & PHYSICAL CHEMI	CAL	——reciprocal lattice lengths of	
		edges	a,* b,* c*
atomic absorption spectrophoto-			
metry	AAS	——interaxial angles direct lattice	α, β, γ
cation-exchange capacity	c.e.c.	———reciprocal lattice	$\alpha^*, \beta^*, \gamma^*$
concentrated	conc.	OPTICAL	
differential thermal analysis	DTA	dispersion, e.g.	r > v
dilute	dil.	transmission electron microscopy	TEM
disintegrations per minute	d.p.m.	extinction angle, e.g	y:c
equivalent U ₃ O ₈	eU_3O_8	infrared	IR
ethylenediamintetra-acetic acid	EDTA	optic axial angle	2V
fugacity	f		O.A.P.
gas chromatography	GC		refr. ind.
heat of formation (absolute tem-			
perature subscript)	ΔHf	refractive indices	n
hydrogen ion conc. acidity	pH		
insoluble residue	insol. res.	of uniaxial mineral	ω, ϵ
isotopes, e.g	⁴⁰ Ar, ⁴⁰ K	of biaxial mineral	α, β, γ
large ion lithophile	LIL	scanning electron microscopy	SEM
loss on ignition	ign. loss	sign of biaxiality	
mid-ocean ridge basalt	MORB	negative	$2V_{\alpha}$ or -
milliequivalent	me.	positive	$2V_{v}$ or +
mass spectrometry	MS	ultraviolet	UV
microgramme	μg	PHYSICAL	
million-years	m.y.	calculated	1000
neutron activation analysis	NAA		calc.
not determined	n.d.	cycles per second	c/s
not found	nt. fd.	degree centigrade	°C
not present	nil	density	D (quote units)
nuclear magnetic resonance	NMR	—, relative, e.g.	D_4^{20}
parts per billion	ppb	electron paramagnetic resonance	e.p.r.
parts per million	ppm	hardness	H.
rare earths	REE	kilobar (0-1 GPa)	kbar
standard mean ocean water	SMOW	melting-point	m.p.
strength of solution, normal	N	micron (10 ⁻³ mm)	μm
———molar	M	nanometre (10 ⁻⁶ mm)	nm
substances in ionic state	***	natural remanent magnetization	n.r.m.
anions, e.g	Cl ⁻ , SO ₄ ²⁻	pressure	P
cations, e.g	K ⁺ , Fe ³⁺	soluble	sol.
thermogravimetric analysis	TGA	specific gravity, terms of reference	
trace	tr.	not known	sp. gr.
X-ray powder diffraction	XRD	temperature	T
X-ray fluorescence analysis	XRF	thermoluminescence	TL
12 Tay hadrescence analysis	AKF	Vickers hardness number	VHN
CRYSTALLOGRAPHIC & STRUCT	URAL	wavelength	λ
8	p	CVA POLIC	
Angstrom unit (10 ⁻⁶ cm)	A	SYMBOLS	
crystal axes	a,b,c	approximately equal to	~
—face indices	(hkl)	equal to	
—form indices	{hkl}	equal to or greater than	≥
—zone indices	[hkl]	equal to or less than	€
indices of X-ray diffractions	hkl	greater than	>
intensity	I	less than	<
—relative	I/I_0	not equal to	<i>≠</i>
interplanar spacing	d	parallel to	.11
mica structural polymorphs	1 M ₁ , 2 M	per cent	∭ '%
Siegbahn units	kX	per mille	%0
unit cell, formula units	\boldsymbol{Z}	nernendicular to	
repeat distances	a, b, c	proportional to	1
		proposition to	∞

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